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(b) Calculate the pH of a solution of 5.0 10<sup>-4</sup> M HCl. (c) Write out the acid dissociation reaction for sodium hydroxide. (d) Calculate the pH of a solution of 7.0 10<sup>-5</sup> M NaOH. Water chapter 2 Lehninger Principles of Biochemistry 6th Edition Nelson Solutions Manualc02Water.qxd 12/6/12 4:12 PM Page S-14

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